## Note

# Remote Bactericidal Effect of Anatase TiO<sub>2</sub> Photocatalytic Nanoparticles Annealed with Low-Temperature O<sub>2</sub> Plasma

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The remote bactericidal effect of TiO<sub>2</sub> photocatalyst, i.e., the bactericidal effect away from the photocatalyst, was successfully achieved using a humidified airflow. The TiO<sub>2</sub> photocatalyst used was anatase-type TiO<sub>2</sub> nanoparticles (NPs) annealed with a low-temperature O<sub>2</sub> plasma. For comparison, anatase-type TiO<sub>2</sub> NPs annealed in the air were used. The bacteria, *Bacillus subtilis*, were placed away from the TiO<sub>2</sub> NPs. The plasma-assisted-annealed TiO<sub>2</sub> NPs significantly inactivated 99% of the bacterial cells in 5 h, whereas the pristine and air-annealed TiO<sub>2</sub> NPs inactivated 88-90% of the bacterial cells. The remote bactericidal effect of plasma-assisted-annealed TiO<sub>2</sub> NPs would be attributed to a larger amount of H<sub>2</sub>O<sub>2</sub> molecules traveled by the airflow from the TiO<sub>2</sub> NPs. The molecules were generated by chemically reacting more photoexcited carriers on the TiO<sub>2</sub> surface with H<sub>2</sub>O and O<sub>2</sub> in the airflow. These photoexcited carriers originated from more oxygen-based species adsorbed and more oxygen vacancies introduced on the TiO<sub>2</sub> surface by the plasma-assisted-annealing.

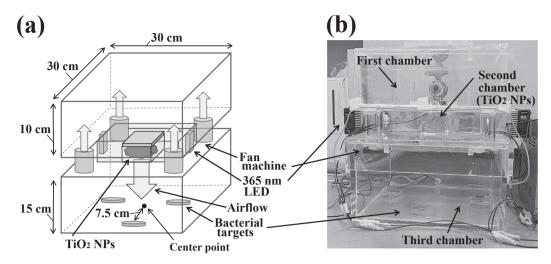
Key words: Remote bactericidal effect / Anatase titanium dioxide nanoparticles / Plasma-assisted annealing / Reactive oxygen species / Hydrogen peroxide.

Superior technologies to control the spread of the coronavirus disease 2019 pandemic have been strongly desired (Khan et al., 2022; Prakash et al., 2022; Pasquale et al., 2021). The use of TiO<sub>2</sub>-based photocatalytic semiconductors is one of the excellent technologies for pathogen disinfection (Khan et al., 2022; Prakash et al., 2022; Pasquale et al., 2021) and antibacterial applications (Singh et al., 2022; Liao et al., 2021; He et al., 2021a; He et al., 2021b; Liu et al., 2021). This reason is that TiO2 irradiated with ultraviolet (UV) is more effective in eliciting microbial death than other photocatalytic semiconductors (Kubacka et al., 2014). The inactivation of bacteria due to TiO<sub>2</sub> is thought to be attributed to reactive oxygen species (ROS) generated by the TiO<sub>2</sub> surface under UV irradiation. Specifically, the electrons and holes photogene-

However, there is a crucial issue regarding the practical use of inactivating bacteria with TiO<sub>2</sub> photocatalysts. In contrast to ionic air purifiers in practice (Kivity et al., 2009; Hanond et al., 2011), TiO<sub>2</sub> photocatalysts hardly inactivate bacteria residing away from TiO<sub>2</sub>. The

rated on the  $TiO_2$  surface react chemically with  $O_2$  and  $H_2O$  in the air to generate  $O_2^-$  and OH radicals, respectively (Arcanjo et al., 2018; Baniamerian et al., 2020; Wu et al., 2020). These ROS would attack the cell wall of bacteria adhering to the  $TiO_2$  surface, damaging the cell membrane and destroying the cellular content (Dalrymple et al., 2010). Therefore, inactivating bacteria with  $TiO_2$  requires bringing bacteria to the  $TiO_2$  surface. For example, the bacterial inactivation of  $TiO_2$  is performed by mixing bacterial suspensions with  $TiO_2$  (Asahara et al., 2009), by dropping the bacterial suspensions onto the  $TiO_2$  surface (Kawakami et al., 2021), and by carrying bacteria to the  $TiO_2$  surface owing to air blowers and fan machines (Xu et al., 2019).

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**FIG. 1**. (a) Schematic drawing and (b) photograph of a remote bactericidal device of TiO<sub>2</sub> NPs using a humid airflow. The developed device consists of three chambers made of transparent acrylic acid resin; from top to bottom, there are the first, second, and third chambers. Three bacterial targets were placed 7.5 cm away from the center point in the third chamber. The TiO<sub>2</sub> NPs immobilized on glass wool were set in the second chamber. The immobilized TiO<sub>2</sub> NPs were irradiated from two directions through a 1-cm think acrylic acid resin using two 365-nm UV LEDs. The humidified air was fed from a humidifier into the first chamber using tap water. A fan machine was used to force the humidified air in the first chamber into the third chamber through the immobilized TiO<sub>2</sub> NPs set in the second chamber. The humidified air reaching the third chamber was returned to the first chamber through four cylinders located near the four corners using four fan machines.

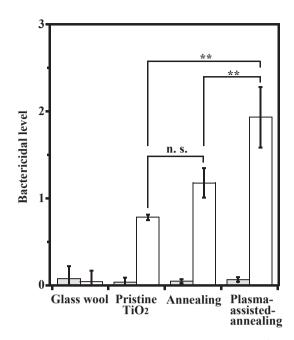
ionic air purifiers release cluster ions or radical ions into the air, inactivating bacteria present away from them or at distant places (Kivity et al., 2009; Hanond et al., 2011). In other words, the ionic air purifiers have remote bactericidal effects. Therefore, new technologies are required to achieve the remote bactericidal effect of  $\text{TiO}_2$  photocatalysts.

In the present study, we have developed a remote bactericidal device of TiO<sub>2</sub> photocatalyst using a humidified airflow, as shown Fig. 1. This device was developed by referencing the literature on photocatalytic remote oxidation (Kubo et al., 2005 and 2006). That study revealed the oxidation effect at distant places but not the remote bactericidal effect (Kubo et al., 2005 and 2006). The TiO<sub>2</sub> photocatalyst used in the present study was anatase-type TiO<sub>2</sub> nanoparticles (NPs) annealed at 300°C with a low-temperature O<sub>2</sub> plasma generated at atmospheric pressure (Kawakami et al., 2020a). The anatase-type TiO<sub>2</sub> NPs were TiO<sub>2</sub> NPs with an average size of 7 nm, called ST-01, purchased from Ishihara Sangyo, Japan. Employing the developed device, we investigated the remote bactericidal effect of plasma-assisted-annealed TiO<sub>2</sub> NPs. For comparison, anatase-type TiO<sub>2</sub> NPs annealed at 400°C in ambient air (Kawakami et al., 2020a) were used. Herein, the remote bactericidal effect of plasma-assisted-annealed TiO<sub>2</sub> NPs is characterized by comparing it with those of the pristine TiO<sub>2</sub> NPs and air-annealed TiO<sub>2</sub> NPs. This remote bactericidal effect is also featured by comparing it with those of ionic air purifiers. The mechanism of remote bactericidal effect is discussed in terms of  $\text{TiO}_2$ -produced ROS in the humidified airflow. The present study emphasizes that the remote bactericidal device of anatase  $\text{TiO}_2$  photocatalytic NPs developed is a valuable technology of inactivating bacteria at distant places.

The developed remote bactericidal device consists of three chambers made of transparent acrylic acid resin; as shown in Fig. 1, from top to bottom, there are the first, second, and third chambers. Three bacterial targets were placed 7.5 cm away from the center point in the third chamber with a volume of 30×30×15 cm<sup>3</sup>. The TiO<sub>2</sub> NPs immobilized on 9-µm diameter glass wool were set in the second chamber with a volume of 5×5×5 cm<sup>3</sup>. For this immobilization, the surface of glass wool 2 g in weight was irradiated for 3 min with an Ar plasma jet to enhance the wettability (Kawakami et al., 2020b). The Ar plasma jet was generated with a twisted wirescylindrical electrode configuration (Kawakami et al., 2020b). After plasma irradiation, a 20-mL TiO<sub>2</sub> dispersion solution containing 0.66-g TiO<sub>2</sub> NPs was dripped onto the plasma-irradiated surface of glass wool. The dipped TiO<sub>2</sub> NPs were then allowed to dry naturally for three days to immobilize the TiO2 NPs. The immobilized TiO<sub>2</sub> NPs were irradiated from two directions through a 1-cm think acrylic acid resin with an area of 5×5 cm<sup>2</sup>

using two 365-nm ultraviolet (UV) light-emitting devices (LEDs) as shown in Fig. 1. The UV intensity at the immobilized TiO<sub>2</sub> NPs was 16.8 mW/cm<sup>2</sup> as measured with an optical power meter (Ophir Nova II). The humidified air was fed from a humidifier into the first chamber using tap water. The temperature and relative humidity (RH) in the chamber were the room temperature and 80%-95% RH, respectively. A fan machine was used to force the humidified air in the first chamber into the third chamber through the immobilized TiO2 NPs set in the second chamber. The speed of airflow generated by the fan machine was 1.15 m/s as measured with an anemometer. The humidified air reaching the third chamber was returned to the first chamber through four cylinders, 2.4 cm in inner diameter and 5 cm long, located near the four corners using four fan machines as shown in Fig. 1. In other words, the humidified airflow was set to be circulated from the first chamber to the third chamber through the second chamber. The air circulation in the developed device, therefore, belongs to the internal air circulation.

The bacteria strain used were Bacillus subtilis (B. subtilis) American Type Culture Collection 6633 (Hassen et al., 2000). The bacterial cells were grown in Luria-Bertani (LB) broth (Nacalai Tesque Lennox, Japan) at 37°C with shaking at 150 rpm for 17 h. The grown bacterial suspensions were centrifuged at 6,500  $\times g$  for 3 min at 4°C min to remove the LB broth and washed twice with sterile ion-exchanged water. The bacterial suspensions of 2×10<sup>8</sup> colony-forming units per mL (CFU/mL) were prepared in sterile ionexchanged water. A 10 µL of the bacterial suspension was dropped onto a surface of 0.8% NaCl agar (SCA) in a dish 6 cm in diameter and 1.4 cm in height. The three bacteria-contained dishes were prepared and placed as the bacterial targets, as shown in Fig. 1. The remote bactericidal experiment was then performed for a time of 5 h. After the remote bactericidal experiment, each of the SCA surface was washed in a 1-mL solution of soybean-casein digest lecithin polysorbate (SCDLP) broth for 1 min to extract the bacteria from the SCA surface. A 0.1-mL broth suspension containing the bacteria was subjected to 10-fold serial dilutions using a 0.9-mL SCDLP broth. The serial dilutions plated on agar plates of SCDLP were incubated at 37°C for 20 h. After incubation, the log numbers of bacterial colonies incubated on the agar plate were assessed using  $|\log (N_t/N_0)|$ , defined as the bactericidal level, where  $N_t$ and  $N_0$  are the colony counts after and before the remote bactericidal experiment, respectively (Jang et al., 2016; Shi et al., 2019). The bactericidal level was measured three times for each TiO2 sample set in the developed device. The measured data were analyzed statistically using the Tukey-Kramer test (Esumi Excel



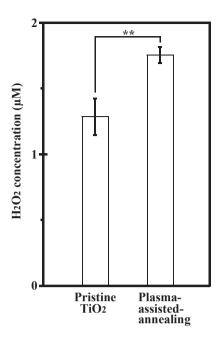
**FIG. 2**. Remote bactericidal effects of glass wool (without  $TiO_2$  NPs), pristine  $TiO_2$  NPs, air-annealed  $TiO_2$  NPs, and plasma-assisted-annealed  $TiO_2$  NPs in the presence (open bars) and absence (shaded bars) of UV irradiation. The symbol, \*\*, corresponds to a *P* value < 0.01, as estimated by the Tukey-Kramer test. The abbreviation, n. s., signifies no significant differences, as estimated by the same test.

Tokei ver. 7.0, Japan). The results are presented as the averaged values with the standard deviations.

Figure 2 shows the remote bacterial effects of the TiO<sub>2</sub> samples set in the developed device. A high bactericidal level signifies a high remote bactericidal effect. The figure also shows the results obtained in the case of glass wool without TiO<sub>2</sub> NPs. In the case of glass wool without TiO<sub>2</sub> NPs, the bactericidal level is below 0.1 in the presence or absence of UV irradiation. This result indicates that the glass wool has no remote bactericidal effect. In the absence of UV irradiation, the bactericidal levels of any TiO<sub>2</sub> samples are below 0.1, indicating that the remote bactericidal effect of TiO2 NPs does not work without UV irradiation. In contrast, the remote bactericidal effects of TiO<sub>2</sub> samples are exerted in the presence of UV irradiation. The bactericidal level of plasma-assistedannealed TiO2 NPs is approximately 2, suggesting a 99% viability reduction or 99% inactivation degree. The bactericidal level of air-annealed TiO<sub>2</sub> NPs, approximately 1, seems higher than that of pristine TiO<sub>2</sub> NPs, approximately 0.8, but there is no statistical difference between the two results. These values are half that of plasma-assisted-annealed TiO<sub>2</sub> NPs. The result suggests that the pristine TiO<sub>2</sub> NPs and air-annealed TiO<sub>2</sub> NPs have 88-90% inactivation degrees. Thus, the plasmaassisted-annealing contributes to increasing the remote bactericidal effect of TiO<sub>2</sub> NPs. On the other hand, the air-annealing does not increase the remote bactericidal effect.

The remote bactericidal effects obtained in the present study are compared with those induced by the ionic air purifiers or plasma cluster ion generators (Digel et al., 2005; Comini et al., 2021). Cluster ions generated with an ionic air purifier (Sharp Corp. Plasma Cluster, Japan) reduced the relative viable number of B. subtilis from 1.0 to 0.2 in 8 h (Digel et al., 2005). This result suggests that the inactivation degree of the generated cluster ions is 80%. Air ions generated with another ionic air purifier (Denso Thermal Systems Plasma Cluster Ionizer, Italy) inactivated 60% of Escherichia coli and 95% of Staphylococcus aureus in 8 h (Comini et al., 2021). These inactivation degrees are lower than that exerted by the remote bactericidal device of plasma-assisted-annealed TiO<sub>2</sub> NPs. These comparisons indicate that the developed remote bactericidal device with the plasma-assisted-annealed TiO2 NPs is an effective means of inactivating bacteria at distant places.

The main factor of the remote bactericidal effect exerted by the plasma-assisted-annealed TiO<sub>2</sub> NPs would be the ROS traveled by the humidified airflow from the TiO<sub>2</sub> NPs. Since the distance from the TiO<sub>2</sub> NPs to the bacterial targets is approximately 16-20 cm and the humidified airflow speed is 1.15 m/s as describe above, the flight time of ROS traveled by the airflow to the bacterial target is estimated to be 140-173 ms. Considering the lifetimes of ROS (Zang et al., 1992; Hayyan et al., 2016), the lifetime of H<sub>2</sub>O<sub>2</sub> molecules, 10<sup>4</sup> s (Zang et al., 1992), and the lifetime of  $O_2^-$  radicals, 1.25 s (Hayyan et al., 2016), are longer than the estimated flight time of ROS traveled by the airflow. According to the literature (Vernez et al., 2017), the ROS predominantly generated on anatase-type TiO2 are  $\mathrm{H_2O_2}$  molecules, whereas  $\mathrm{O_2}^-$  radicals are generated mainly on rutile-type TiO<sub>2</sub>. Therefore, only H<sub>2</sub>O<sub>2</sub> molecules in the humidified airflow would be allowed to reach the bacterial targets because the anatase-type TiO<sub>2</sub> NPs were used in the developed device. Figure 3 shows the concentrations of H<sub>2</sub>O<sub>2</sub> molecules, which were generated by the pristine TiO<sub>2</sub> NPs and plasmaassisted-annealed TiO<sub>2</sub> NPs in a 5-h UV exposure and then traveled to the bacterial targets by the airflow in the remote bactericidal device. The H<sub>2</sub>O<sub>2</sub> concentration was measured by the colorimetric method, i.e., the absorptiometry at a wavelength of 560 nm, based on the peroxide-mediated oxidation of Fe<sup>2+</sup> followed by the reaction of Fe<sup>3+</sup> with xylenol orange (Jiang et al., 1990; Shirai et al., 2022). The H<sub>2</sub>O<sub>2</sub> measurement was performed three times for each TiO<sub>2</sub> sample under the same experimental conditions as those used for the



**FIG. 3**. Concentrations of  $H_2O_2$  molecules, generated by pristine  $TiO_2$  NPs and plasma-assisted-annealed  $TiO_2$  NPs in a 5 h-UV exposure and then traveled to the bacterial targets in the remote bactericidal device. The two results have a significant difference of \*\*P < 0.01 as estimated by the two-tailed and unpaired t-test. The  $H_2O_2$  concentration was measured three times for each  $TiO_2$  sample.

remote bactericidal experiment. In the case of glass wool, the measured H<sub>2</sub>O<sub>2</sub> concentration was 0.6-0.7 µM in the presence and absence of UV exposure (data not shown). The occurrence of this H<sub>2</sub>O<sub>2</sub> concentration may be due to the presence of iron oxides such as Fe (OH)<sub>2</sub> contained in the humidified air produced from tap water because the glass wool does not inactivate the bacteria (Fig. 2). In the absorptiometry for H<sub>2</sub>O<sub>2</sub>, the presence of iron oxides may increase the absorbance, seemingly increasing the H<sub>2</sub>O<sub>2</sub> concentration. Therefore, this increase is thought not to be a net increase in the H<sub>2</sub>O<sub>2</sub> concentration. In the absence of UV exposure, the H<sub>2</sub>O<sub>2</sub> concentration measured in each TiO<sub>2</sub> sample was similar to that observed for glass wool. The similarity suggests that the occurrence of this H<sub>2</sub>O<sub>2</sub> concentration may originate from the presence of iron oxides in the humidified air because each TiO<sub>2</sub> sample in the absence of UV exposure does not inactivate the bacteria, as in the case of glass wool. As shown in Fig. 3, the  $H_2O_2$ concentration of plasma-assisted-annealed TiO2 NPs is approximately twice as high as that of pristine TiO<sub>2</sub> NPs. This higher H<sub>2</sub>O<sub>2</sub> concentration would be attributed to the increased effective density of photoexcited carriers, which originates from more oxygen-based species adsorbed and more oxygen vacancies introduced on the TiO<sub>2</sub> surface by the plasma-assisted-annealing

(Kawakami et al., 2020a). The adsorbed oxygen-based species and introduced oxygen vacancies facilitate the charge separation by increasing the depletion layer width and trapping the photogenerated electrons, thus increasing the effective density of photoexcited carriers (Kawakami et al., 2020a). These photoexcited carriers would react chemically with H2O and O2 in the humidified airflow, thus generating more H<sub>2</sub>O<sub>2</sub> molecules (He et al., 2021a). Specifically, more photogenerated holes react with H<sub>2</sub>O in the humidified airflow, generating more OH radicals and H<sup>+</sup> ions. More photogenerated electrons react with O<sub>2</sub> in the airflow, generating more O<sub>2</sub> radicals. The generated O<sub>2</sub> radicals react with the generated H<sup>+</sup> ions to produce more HO<sub>2</sub> radicals. As a result, more H<sub>2</sub>O<sub>2</sub> molecules are generated through 2  $HO_2$  radicals  $\rightarrow H_2O_2+O_2$ . The  $H_2O_2$  concentration result correlates with the remote bactericidal effect (Fig. 2). Thus, more H<sub>2</sub>O<sub>2</sub> molecules traveled by the airflow would contribute primarily to the remote bactericidal effect of plasma-assisted-annealed TiO<sub>2</sub> NPs.

We have developed the remote bactericidal device of anatase TiO2 photocatalytic NPs using the humidified airflow. Using this device, we clarified the remote bactericidal effect of plasma-assisted-annealed TiO<sub>2</sub> NPs by comparing it with those elicited by the pristine TiO<sub>2</sub> NPs and by the TiO<sub>2</sub> NPs annealed in ambient air. The plasma-assisted-annealed TiO2 NPs significantly inactivated 99% of the bacterial cells in 5 h, whereas the pristine and air-annealed TiO<sub>2</sub> NPs inactivated 88-90% of the bacterial cells. This remote bactericidal effect was superior to those of the ionic air purifiers. The remote bactericidal effect of plasma-assisted-annealed TiO<sub>2</sub> NPs would be attributed to the larger amount of H<sub>2</sub>O<sub>2</sub> molecules traveled by the humidified airflow from the TiO<sub>2</sub> NPs. These H<sub>2</sub>O<sub>2</sub> molecules were generated through the chemical reactions of more photoexcited carriers on the TiO2 surface with H2O and O2 in the humidified airflow. These photoexcited carriers originated from more oxygen-based species adsorbed and more oxygen vacancies introduced on the TiO2 surface by the plasma-assisted-annealing. The findings obtained are vitally essential for fully understanding the remote bactericidal effect of photocatalysts. These findings will also provide a new perspective on photocatalytic applications for inactivating bacteria.

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### **CONFLICT OF INTEREST**

The authors declare no conflict of interest.

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